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pH - Wikipedia

In chemistry, pH (p H e t) is a scale used to specify how acidic or basic a water-based solution is. Acidic solutions have a lower pH, while basic solutions have a higher pH. Die Theorie der quantitativen Bestimmung der Acidit t und ...

Die Titrationsacidit t, welche man noch trennen kann in Neutralkapazit t und quivalenkapazit t, 2, die Wasserstoffzahl bzw. p h, 3, die Nachgiebigkeit bzw. Pufferung, Pufferung. This is a preview of subscription

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What is the amount of caustic need to increase the pH of ...

What is the amount of caustic need to increase the pH of 1L water from 7 to 9 and the amount of HCL to lower the pH of 1 L water from 7 to 5 ?

S. P. L. Sørensen - Wikipedia

Søren Peter Lauritz Sørensen (9 January 1868–12 February 1939) was a Danish chemist, famous for the introduction of the concept of pH, a scale for measuring acidity and alkalinity.

Measurement of pH of the Skin Surface: I. Technique ...

Full Length Report 1 2 1 From the Department of Dermatology and Syphilology, Harvard Medical School and Massachusetts General Hospital; C. Guy Lane, M.D., Chief. 2 This investigation was aided by a grant for research from the National Oil Products Company, Harrison, New Jersey.

The cleavage reaction of trans-aquoerythrochromium(III) ...

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Klinische und radiologische Beobachtungen an ruminierenden ...

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